

Student Scholar Symposium Abstracts and Posters

Center for Undergraduate Excellence

Spring 5-1-2024

Synthesis of Thermoresponsive Poly(N-isopropyl acrylamide) Based Core-Shell and Hollow Shell Nanogel with Tunable Core and Shell Thickness

Mohamad Hijazi Chapman University, mhijazi@chapman.edu

Molla R. Islam Chapman University, islam@chapman.edu

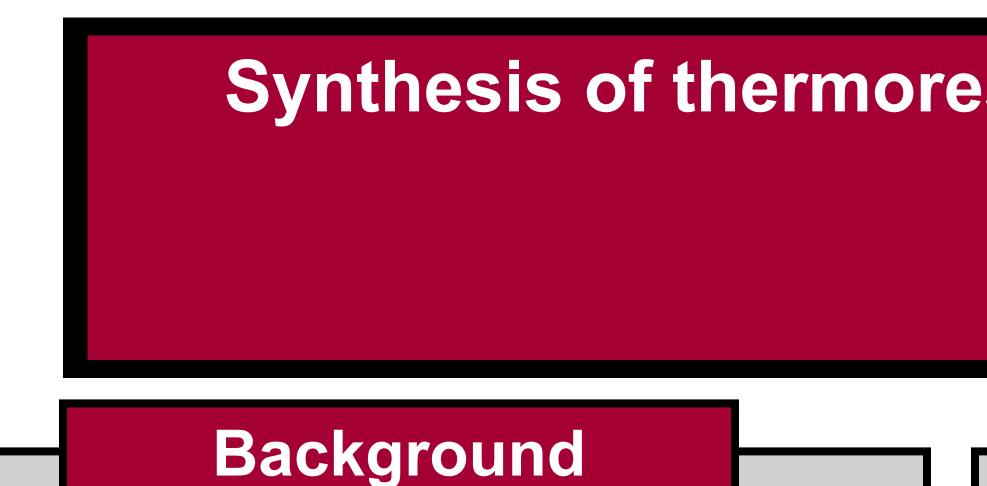
Follow this and additional works at: https://digitalcommons.chapman.edu/cusrd_abstracts

Part of the Medicinal and Pharmaceutical Chemistry Commons, Medicinal-Pharmaceutical Chemistry Commons, Other Pharmacy and Pharmaceutical Sciences Commons, and the Pharmaceutics and Drug Design Commons

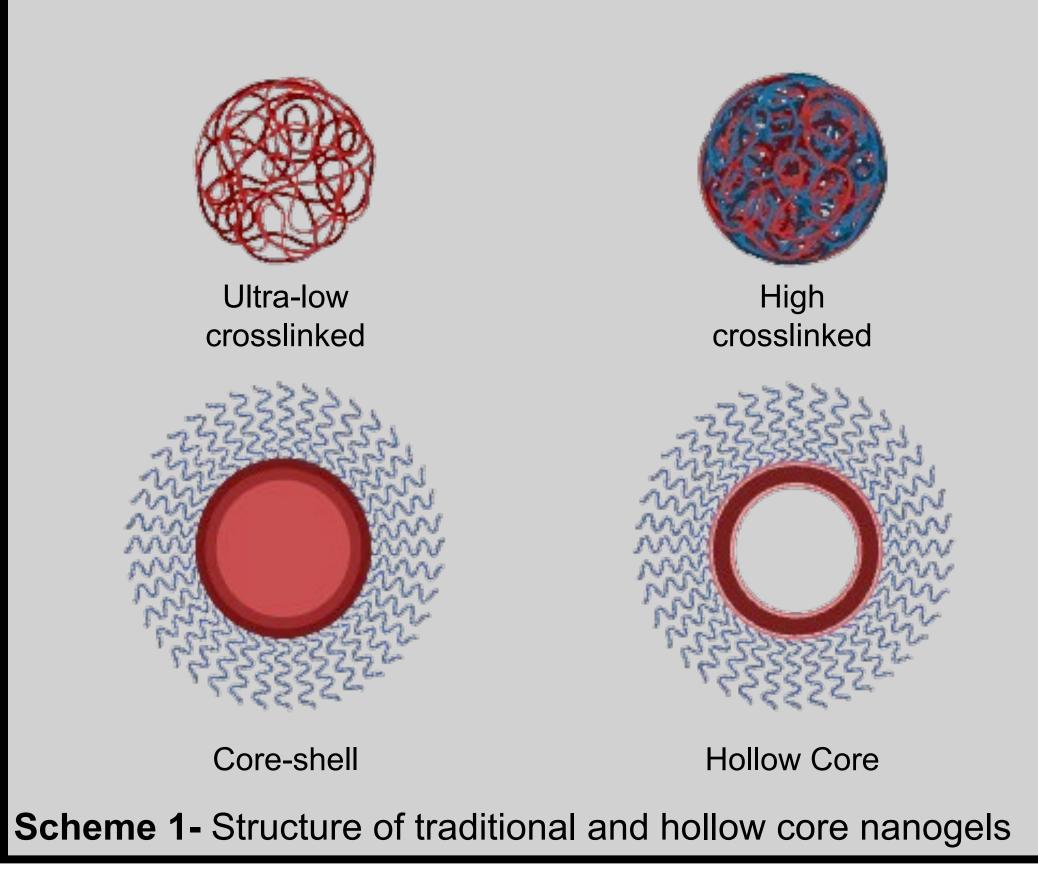
Recommended Citation

Hijazi, Mohamad and Islam, Molla R., "Synthesis of Thermoresponsive Poly(N-isopropyl acrylamide) Based Core-Shell and Hollow Shell Nanogel with Tunable Core and Shell Thickness" (2024). *Student Scholar Symposium Abstracts and Posters*. 645. https://digitalcommons.chapman.edu/cusrd_abstracts/645

This Poster is brought to you for free and open access by the Center for Undergraduate Excellence at Chapman University Digital Commons. It has been accepted for inclusion in Student Scholar Symposium Abstracts and Posters by an authorized administrator of Chapman University Digital Commons. For more information, please contact laughtin@chapman.edu.



- three-dimensional Nanogels are polymer networks adaptable in size, shape, and softness.^{1,2}
- Can be made responsive to external factors like temperature and pH.¹
- Traditional types (e.g., ultra-low crosslinked, high crosslinked, coreshell) are used in drug delivery but lack protective compartments.^{3,4}
- Hollow core nanogels offer protective storage for drugs, featuring a hollow center.⁵
- Control size and diameter can be manipulated by changing reaction conditions.⁶

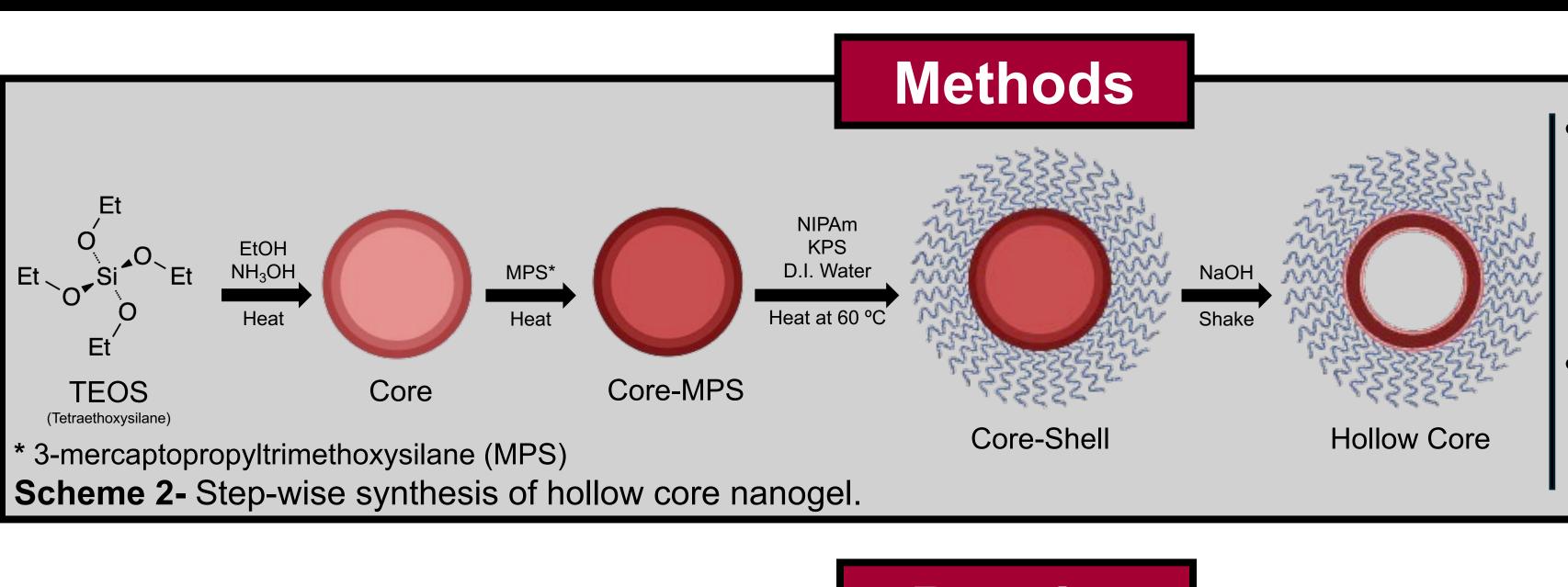


Research Objectives

- Investigate the effect of temperature to control the core particles precursor of hollow-shell nanogels.
- Wrapping the core with ultra-low crosslinked NIPAm shells to improve drug-loading efficiency.
- Dissolve TEOS-based silica cores within core-shell particles to form hollow-core with a precise thickness of the shell.

Synthesis of thermoresponsive poly(N-isopropyl acrylamide) based core-shell and hollow shell nanogel with tunable core and shell thickness

Mohamad Hijazi and Molla Islam Schmid College of Science and Technology, Chapman University





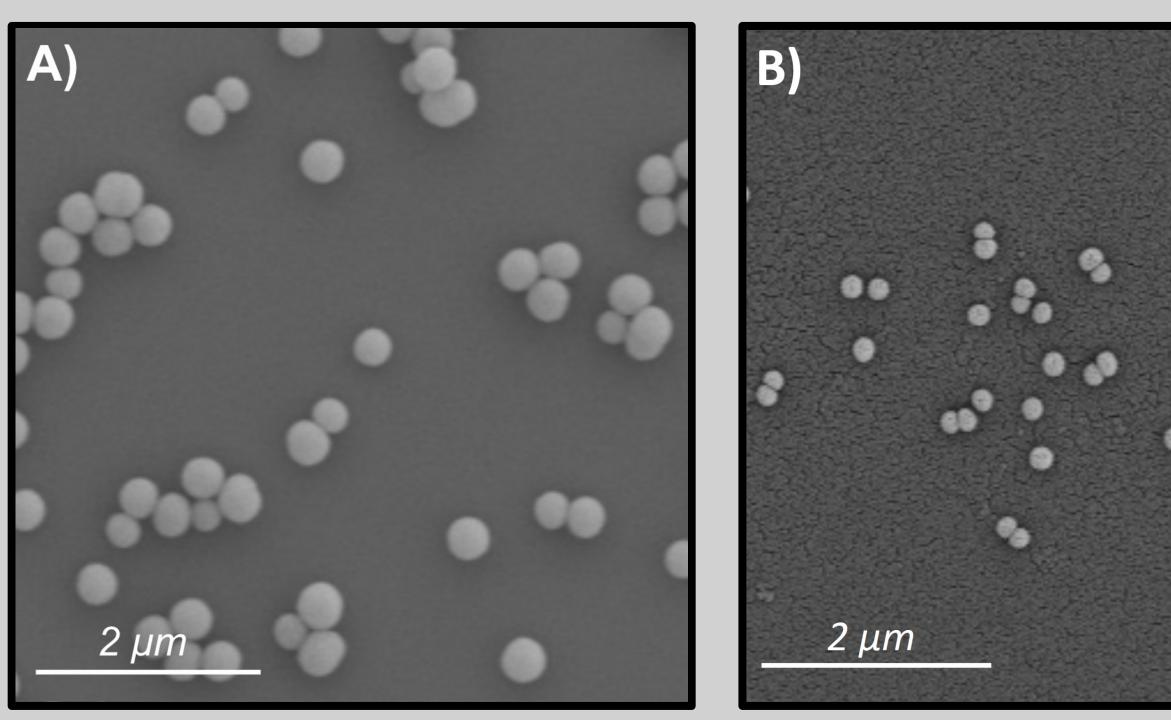


Figure 1. Silica core nanogels were imaged using a Scanning Electron Microscope (SEM) at 30 °C (A), 60 °C (B), and core-shell nanogels using Optical Microscopy (OM) (C).

- SEM images reveal uniformly sized spherical particles synthesized at both 30°C and 60°C, with those synthesized at 30°C appearing larger.
- OM images show core-shell structures characterized by a white halo surrounding a dense core.

Synthesis	Hydrodynamic Radius (nm) (20 °C)			S
Temperature	CORE	CORE-SHELL		1
30 °C	195 ± 10	449 ± 20		S
60 °C	66 ± 3	233 ± 10	•	C
Table 1 Average hydrodynamic radius measured with				h

Table 1. Average hydrodynamic radius measured with Dynamic Light Scattering (DLS) of core and core-shell of particles synthesized at 30 and 60 °C.

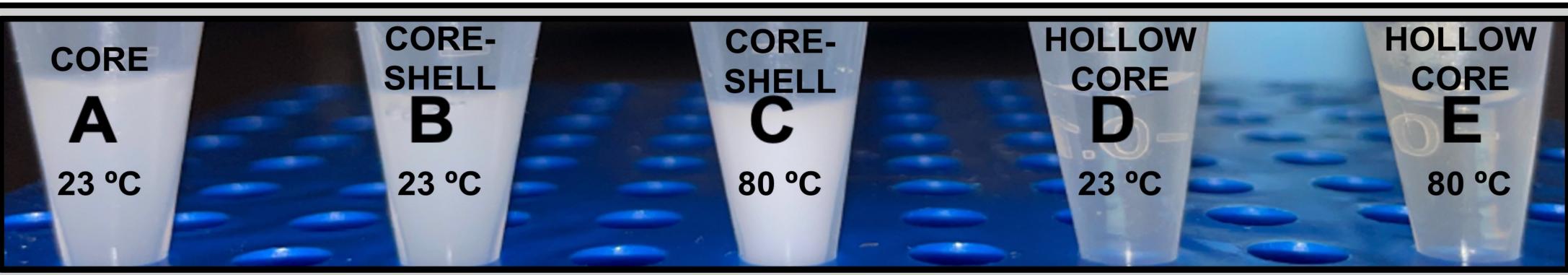


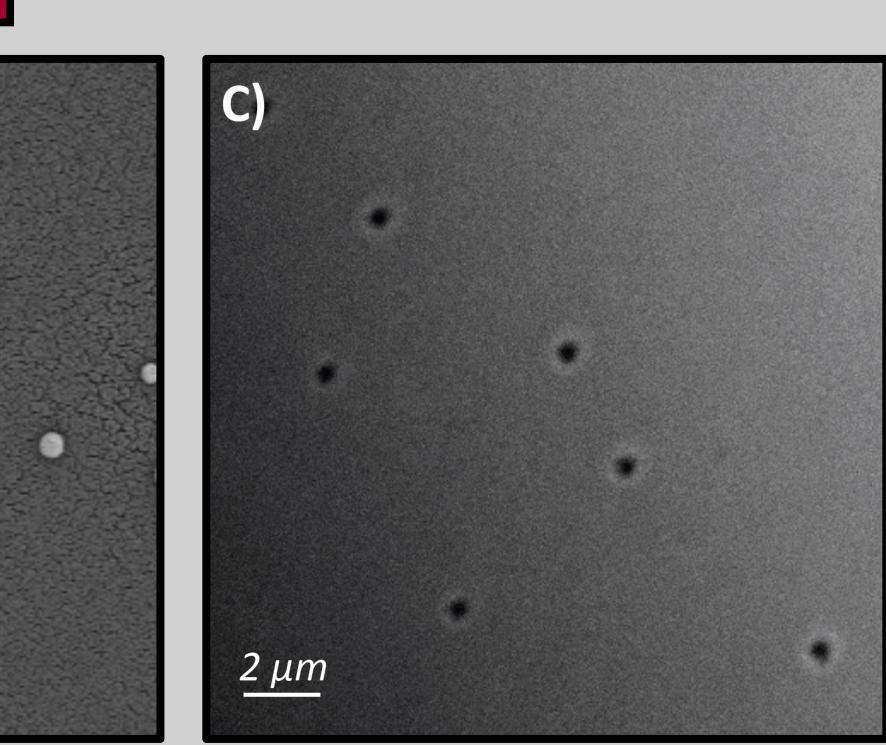
Figure 2. Photographs of silica core at 23°C (A), core-shell at 23°C (B) and 80°C (C), hollow core at 23°C (D), and hollow core at 80°C (E).

- Morphological changes in core-shell particles are evident when heated from 23°C to 80°C, where they shift to a darker white hue at 80°C.
- The hollow core solution, initially clear, transforms into a pearlescent clear color with a slight spectral reflection at 80°C, demonstrating the thermoresponsiveness of the NIPAm shell.

SEM and OM were used for particle characterization. DLS was used for

Size

measurements.



ilica core particles synthesized at 30°C measure 95 nm, whereas those synthesized at 60°C are maller at 66 nm.

ore-shell particles synthesized at 30°C exhibit a ydrodynamic radius of 449 nm, while those at 60°C have a smaller radius of 233 nm.

- particles due to slower nucleation and polymer growth.⁷
- Lower temperatures led to larger
- Optical microscope images showed less dense NIPAm shells compared to the silica core.⁸
- Color changes upon heating indicated NiPAM shell's
- thermoresponsive behavior.⁸
- High temperatures caused NIPAm
- shell contraction, darkening the solution.⁸

- for

Conclusion

 Particle shape and size distribution remained uniform despite synthesis temperature changes.⁶

Future Work

 Study drug loading capacity with model drugs like Fluorescein. Monitor drug release kinetics for timing and efficiency. Manipulate nanogel responsiveness enhanced control over drug release and targeting.

Acknowledgements

Special thanks to Chapman's Schmid College of Science and Technology, Chapman's Office of Research, the High-Resolution Imaging Facility for their support, and the BIO 494 course.

References

